

EXCEPTIONAL CAPABILITY FOR Pb2+ UPTAKE FROM AQUEOUS MEDIA A Zr4+ MOF WITH DITHIOCARBAMATE FUNCTIONALITY SHOWING



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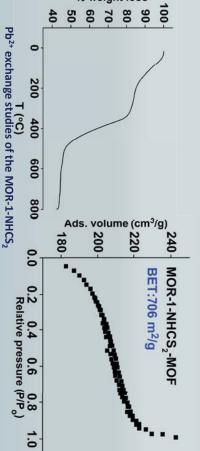
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efficiency to remove heavy metal ions from aqueous media Heavy metal ions represent major contaminants for the water resources, and, especially, the pollution from Pb²⁺ is a major concern, given that Pb²⁺ can cause acute and chronic poisoning effects and targets the majority of the organs in the human body. For these reasons, the elimination of lead from water systems is considered highly important in current environmental research. Recently, we reported the material $[Zr_sO_4(OH)_4(NH_3^+-BDC)_6]Cl_6$ (MOR-1), a MOF which showed exceptional capability to selectively absorb hexavalent chromium and other anionic species. Herein, we present a post-synthetic modification of MOR-1 with CS₂, aiming in a sorbent (MOR-1-NHCS₂) with high

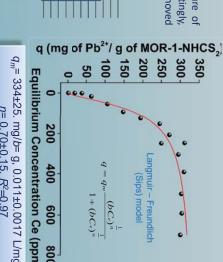
Characterization of MOR-1-NHCS,



% weight loss



Pb $^{2+}$ by MOR-1-NHCS $_2$ was remarkably fast. Interestingly, 95.6% of the initial Pb $^{2+}$ content (c_i=1ppm) was removed within only 1 min of MOR-1-NHCS $_2$ /solution contact. The kinetic investigations indicated that the capture of Kinetic Study



in the

competitive cationic species, the lead removal

presence of high excess of several

determined in

was found high and similar

with

three

bottled

water

genuine water

samples, we performed tests

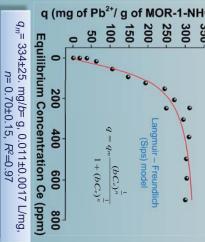
NHCS₂-MOF is suitable for Pb²⁺ sorption

investigate whether MOR-1-

Selectivity Study

intentionally contaminated with traces of Pb2+

(initial total Pb^{2+} concentration = 1 ppm). Even



% sorption of Pb2+

98

96

4

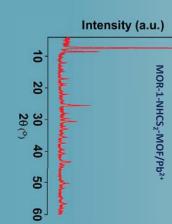
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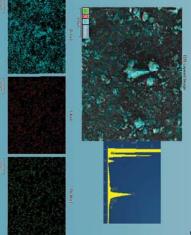
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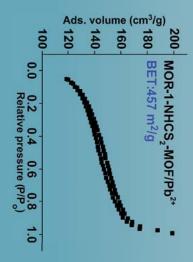
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Characterization of Pb2+-loaded MOR-1-NHCS,







Acknowledgments

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